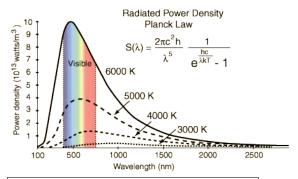
Why and when do we need quantum mechanics?

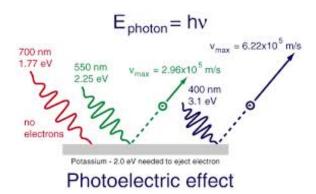
- Many observed phenomena cannot be descibed by classical physics
- QM is applicable when the action is of the order of Plank's constant
- QM is most often necessary to describe the microworld (though some macroscopic phenomena require quantum mechanics as well)

• Max Planck (1901). Black body radiation. Idea of quantized energy, or quants: $E = nh\nu$

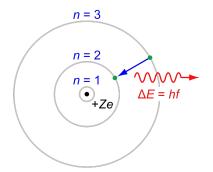


$$h = 6.62 \times 10^{34} \text{ J} \cdot \text{s} - \text{Planck's constant}$$

• Albert Einstein (1905). Photoelectric effect. $h\nu = W + \text{K.E.}$



• Niels Bohr (1913). Hydrogen atom model. $\mu vr \equiv L = n\hbar$



$$E_n = -\frac{\mu e^4}{2\hbar^2} \frac{Z^2}{n^2}$$

Sommerfeld (1915). Extended Bohrs model to elliptical orbits.

Louis de Broglie (1923). Particle-wave dualism.

$$p=\frac{h}{\lambda}$$
, $E=h\nu$

$$\mathbf{p}=\hbar\mathbf{k}$$

• Werner Heisenberg (1925). Matrix mechanics. Uncertainty principle $\Delta x \Delta p_x \geq \hbar$

• Erwin Schrödinger (1926). Wave mechanics. The Schrödinger equation.

$$i\hbar \frac{\partial \psi}{\partial t} = H\psi$$

 Paul Dirac (1927). Shows equivalence of the matrix and wave mechanics.